Course Specifications

Programme(s) on which the course is given **M**. **Sc.** Major or Minor element of programmers **Major** Department offering the programmer: **Chemistry** Department offering the course: **Chemistry** Academic year / Level: 2012/2013 Date of specification approval: 2012

A- Basic Information

| Title: Nuclear Chemistry | | Code: CH6213 |
|--------------------------|----------------|--------------|
| Credit Hours: 2 h | Lecture:2 | |
| Tutorial: 2 | Practical: 0 h | Total: 2 h |

B- Professional Information

1 – Overall Aims of Course

Provide the student with the essential concepts about nuclear chemistry and radiation and their application in all fields (industry – chemistry – agriculture)

2 – Intended Learning Outcomes of Course (ILOs)

a- Knowledge and Understanding:

After completing the course the student should be able to

a1- show the type of nuclear particles.

a2- explain nuclear chemistry and its application.

a3- Take an account on the types of nuclear reactors and their contents.

b- Intellectual Skills

b1- develop the students' knowledge of radiation reactions.

b2- Improve the mean of radiation chemistry and its benefits and harmites.

b3- discuss what has been developed in this field

c- Professional and Practical Skills

c1-apply studied subjects in chemistry

c2- relate applications of tracers with industry

c3- use applications of nuclear chemistry in biology and medicine

c4- use applications of the studied subjects in manufacturing of weapons

c- General and Transferable Skills

- **d1**) Enhancing the writing and oral communication capability of the students.
- d2) Capability of solving the problems.

3- Contents

| Topic | No. of | Lecture | Tutorial/Pra |
|-------------------------------|--------|---------|--------------|
| | hours | | ctical |
| Introduction | 4 | 3 | 2 |
| Nuclear stability | 4 | 3 | 2 |
| Types of reactions | 4 | 3 | 2 |
| Nuclear reactors | 4 | 3 | 2 |
| Application of tracers | 4 | 3 | 2 |
| Detection of radiation | 4 | 3 | 2 |
| Effect of radiation on matter | 4 | 3 | 2 |

4- Teaching and Learning Methods

4.1- lectures.

4.2- illustrated lecture

- 4.3- discussion
- 4.4- assignment

5- Student Assessment Methods

5.1- written examination to assess the understanding and comprehension **5.2-** Oral discussion.

Assessment Schedule

| Assessment 1 short exam (class activities) | Every week |
|---|------------|
| Assessment 2 mid-term (written and practical) | Week 7 |
| Assessment 3 final-term (written and practical) | Week 14 |

Weighting of Assessments

| Mid-Term Examination | 20% |
|------------------------|------|
| Final-term Examination | 60% |
| Semester Work | 20% |
| Total | 100% |

6- List of References

6.1- Text books

i) MODERN NUCLEAR CHEMISTRY, WALTER D. LOVELAND Oregon State University.DAVID J. MORRISSEY Michigan State University. GLENN T. SEABORG University of California, Berkeley (2006).

ii) Inorganic chemistry, ELBS (1994).

7- Facilities Required for Teaching and Learning:

Overhead projector, field visits and video films

Course Coordinator: Dr. Yasser Kamal

Head of Department: Prof. Ahmed Abd-Elmged

Date: 2012